MODELING THE CHEMISTRY OF DENSE INTERSTELLAR CLOUDS. I. OBSERVATIONAL CONSTRAINTS FOR THE CHEMISTRY

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ABSTRACT

A search for correlations arising from molecular line data has been made in order to place constraints on the chemical models of interstellar clouds. Observations of CO, H₂CO, NH₃, HC₃N, HC₅N, and several other species have been compiled. The column density of H₂ has been obtained from determinations of A_V . At 10²¹ H₂ cm⁻², N(CO) for dark clouds is a factor of 6 greater than the value for diffuse clouds. This result implies that the strength of the UV radiation field where CO shields itself from dissociation is approximately one-half the strength of the average Galactic field. Also, the dark cloud data indicate that the abundance of CO continues to increase with A_V for directions with $A_V \leq 4$ mag, although less steeply with N(H₂) than is the case for diffuse clouds. For H₂CO, a quadratic relationship is obtained in plots versus H₂ column density. The data suggest a possible turnover at the highest values for A_V (\geq 4–5 mag). Ammonia shows no correlation with H₂, C¹⁸O, HC₃N, or HC₅N, in part because for dark clouds the column density of NH₃ is nearly constant at 10¹⁵ cm⁻². A strong correlation is found between HC₅N and HC₃N, indicating the chemical link between the cyanopolyynes. Possible trends are also exhibited in plots of the cyanopolyynes versus C₄H, but the weak correspondences with C¹⁸O suggest that emission from the cyanopolyynes and C¹⁸O arises in separate regions.

Subject headings: interstellar: molecules — molecular processes

I. INTRODUCTION

Column densities are the basic observational data in studies of interstellar molecules. Correlations among the observed column densities N(X) of molecular species X are therefore very valuable in constraining theoretical chemical or chemicaldynamical models of interstellar molecular clouds. Equation (1),

$$N(\mathbf{A}) = a \lceil N(\mathbf{B}) \rceil^b , \qquad (1)$$

represents a simple, yet broad-based correlation. The constants a and b in equation (1) are expected to be governed by chemical kinetics and by the physical structure of the cloud. For chemically unrelated species, the exponent helps to determine spatial correspondences within the cloud. For instance, Federman and collaborators (Federman, Danks, and Lambert 1984; Federman and Lambert 1988) argued that a value of 1 for b in the case of A = CH and $B = H_2$ implies that the chemistry of the CH molecule does not involve any neutral precursor species. Given the difficulties involved in the minimization process for nonlinear, least-squares fitting, the constants a and b in equation (1) are derived most easily from a linearized expression:

$$\log N(\mathbf{A}) = \log a + b \log N(\mathbf{B}) . \tag{2}$$

Throughout this paper, therefore, we work in the logarithmic domain and concentrate our efforts on observations of dark clouds. We present plots of log N(X), where X is H₂, CO, H₂CO, NH₃, HC₃N, HC₅N, and C₄H, and values of *a* and *b* whenever physically reasonable.

Correlations among molecular species observed in dark clouds have been sought in the past (e.g., Frerking, Langer and

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Wilson 1982; Benson and Myers 1983; Churchwell, Nash, and Walmsley 1984; Millar and Freeman 1984*a*; Cernicharo and Guelin 1987). Several studies have displayed results on graphs with linear axes, a technique that has disadvantages. Most important, it leads one to seek relationships of the form

$$N(\mathbf{A}) = \alpha + \beta N(\mathbf{B}) . \tag{3}$$

Such relationships do not satisfy the asymptotic boundary condition that $N(A) \rightarrow 0$ when $N(B = H_2) \rightarrow 0$. Allowance for $\alpha \equiv 0$ makes equation (3) a special case of equation (1) with b equal to 1. The present work also builds upon the previous logarithmic analyses by incorporating more data and more species in the analysis.

A discussion of the compiled data appears in the next section (§ II). Our results for oxygen-bearing and nitrogen-bearing species are described in §§ III and IV, respectively. This breakdown is partly due to the fact that our compilations for CO and H₂CO apply to the outer regions ($A_V \leq 4$ mag) of dark clouds, while the results for NH₃, HC₃N, HC₅N, and several carbon-chain molecules apply predominantly to the cores $(A_V \ge 4 \text{ mag})$ of dark clouds. For cloud cores, extinction measurements represent lower limits, and comparisons involving species other than H_2 are more meaningful. Data on $C^{18}O$ are used in analyzing the correlation of CO with H₂. Detailed interpretation of the physical significance of the derived values of a and b are beyond the scope of this empirical paper. However, we make an exception in the case of CO because of the timeliness of our results in light of recent strides made in the modeling of CO self-shielding (e.g., van Dishoeck and Black 1988). The final section, § V, presents our conclusions.

II. THE MOLECULAR DATA

The molecular data for dark clouds are compiled from measurements that are available in the literature. The sources for our compilation are given in Table 1. Several constraints are

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applied to the compilation so that a self-consistent approach is obtained. Foremost, we do not consider data from clouds with substantial amounts of massive star formation, including the ρ Oph Cloud, because of its effects on the chemistry in the cloud. Second, we restrict ourselves to measurements for directions with 1 mag $< A_V < \sim 5$ mag. The lower limit is necessary to ensure that the typical telescope beam $(\sim 2')$ does not inadvertently sample a dense clump near an unreddened sightline; the rough upper limit is required only for correlations between the molecule of interest and A_V [or equivalently $N(H_2)$]. Other comparisons for cores of clouds are possible. Third, several commonly observed species, such as HCN and HCO⁺, are not included in the compilation because of severe saturation in the measured transitions for the main isotopic species. Fourth, not all of the available references are suitable for our study. Only those studies for which a specific value for the parameters of interest [N(X) or $\int T_L dv$, where T_L is the line temperature] is obtainable without interpolation are utilized. Results presented only in the form of contour plots are not used. Although contour maps are quite useful for displaying similarities and differences in molecular distributions, they are less useful for the necessary direct comparisons needed in chemical modeling. Fifth, to obtain reasonable statistics, we analyze those data on a limited number of species that have more than 10 entries.

Fifth, to obtain reasonable statistics, we analyze those data on a limited number of species that have more than 10 entries. Last, we obtain N(X) for a given species at a specific position by averaging different sets of measurements after converting to the same set of excitation parameters. The fourth and last points are discussed in more detail below.

TABLE 1

REFERENCE FOR OBSERVATIONS						
Molecule	References					
Diffuse Clouds ^a						
со	1, 2, 3					
Dark Clouds						
CO H_2 CO NH_3 HC_3N HC_5N C_4H	4, 5, 6, 7, 8 9, 10, 11, 12, 13 14, 15, 16, 17 15, 16, 18, 19, 20 14, 15, 19 15, 21					

^a The UV data are results based on fitting profiles with unresolved structure to the absorption lines (Federman and Lambert 1988).

REFERENCES.-(1) Federman et al. 1980; (2) Tarafdar and Krishna Swamy 1982; (3) Federman and Lambert 1988; (4) Baudry et al. 1981; (5) Myers, Linke, and Benson 1983; (6) Ungerer et al. 1985; (7) Duvert, Cernicharo, and Baudry 1986; (8) Cernicharo and Guelin 1987; (9) Minn and Greenberg 1979; (10) Sherwood and Wilson 1981; (11) Cohen et al. 1983; (12) Pöppel, Rohlfs, and Celnik 1983; (13) Minn and Greenberg 1987; (14) Benson and Myers 1983; (15) Cernicharo, Guelin, and Askne 1984; (16) Churchwell, Nash, Walmsley 1984; (17) Gaida, and Ungerechts, and Winnewisser 1984: (18) Bujarrabal et al. 1981; (19) Tölle et al. 1981; (20) Sorochenko, Tolmachev, and Winnewisser 1984; (21) Guelin, Friburg, and Mezaoui 1982.

The data on C¹⁸O emission and H₂CO absorption predominantly come from maps of several dark clouds. For C¹⁸O, maps of 3C 111 (Ungerer *et al.* 1985), L1495 (Duvert, Cernicharo, and Baudry 1986), and HCL2 (Cernicharo and Guelin 1987) are used. Maps of H₂CO absorption at 6 cm for TMC-2 (Pöppel, Rohlfs, and Celnik 1983), globules near NGC 2264 (Minn and Greenberg 1979), HCL2 (Sherwood and Wilson 1981) and L1709 (Minn and Greenberg 1987) are incorporated into the compilation. In addition to these maps, individual positions, usually the cloud core, have been observed; measurements of C¹⁸O by Baudry *et al.* (1981) and Myers, Linke, and Benson (1983) and of H₂CO by Cohen *et al.* (1983) are included.

For ammonia and the complex molecules, measurements of cloud cores represent the majority of the data compiled. The cores used in the compilation and the molecular results appear in Table 2. The (0, 0) positions for most cores are from Myers, Linke, and Benson (1983). The core positions for TMC-3A and L183S are from Churchwell, Nash, and Walmsley (1984), and the position for HCL2-B is from Cernicharo, Guelin, and Askne (1984). Separate observations within 2' of a given position are considered together in the compilation. If the separate observations are of the same molecule, the resulting column densities are an average for the specific position. Furthermore, maps of TMC-1 (Tölle *et al.* 1981; Guelin, Friberg, and Mezaoui 1982; Gaida, Ungerechts, and Winnewisser 1984) and of TMC-1A and TMC-1C (Cernicharo, Guelin, and Askne 1984) are utilized.

Most of the sources of data have presented results in terms of column densities. The sources listed in Table 1 have derived the C¹⁸O column density in essence under the assumption of LTE. Myers, Linke, and Benson (1983) and Ungerer *et al.* (1985) have improved upon this assumption somewhat by specifically estimating the optical depth and excitation temperature, but as pointed out by Myers, Linke, and Benson, the results are comparable to those of LTE. The formaldehyde data have been made self-consistent by always using an excitation temperature of 1.7 K to obtain $N(H_2CO)$ of the 1₁₁ state (Cohen *et al.* 1983); slightly different values for the excitation temperature would displace the data in the figure by less than 0.1 dex. The column density (in units of cm⁻²) in the 1₁₁ state has been derived from

$$V(H_2CO) = 2.3 \times 10^{13} \text{ W}, \qquad (4)$$

where the equivalent width W equals T_L times Δv (Minn and Greenberg 1979, 1987). The line temperature is T_L , and Δv is the full width at half-intensity of the line. Pöppel, Rohlfs, and Celnik (1983) have tabulated their results in terms of equivalent widths; equation (4) has been used to derive the corresponding column densities.

The column densities for NH₃, HC₃N, HC₅N, and C₄H are based on statistical equilibrium, usually large velocity gradient, models. Gaida, Ungerechts, and Winnewisser (1984), however, obtained the optical depth and excitation for NH₃ emission from profile fitting. For several cloud cores, multiple measurements for NH₃ and HC₃N exist, and the derived column densities spanned a range of at most, and usually less than, a factor of 3. This fact gives us confidence in the values for column density used in our analysis. For C₄H, data of Guelin, Friberg, and Mezaoui (1982) are somewhat arbitrarily scaled to the data of Cernicharo, Guelin, and Askne (1984) because (1) the two data sets revealed similar slopes in plots of $N(C_4H)$ versus either $N(HC_5N)$ or $N(HC_3N)$ but were offset

DATA FOR CLOUD CORES										
	CENTRAL POSITION		OFFSETS ^a	log N(X)						
CLOUD	a(1950)	δ(1950)	$(\Delta \alpha, \Delta \delta)$	N(NH ₃)	N(HC ₃ N)	N(HC ₅ N)	N(C ₄ H)	References ^b		
L1489	04 ^h 01 ^m 45 <u></u> ^s 0	26°10′33″	(0, 0)	15.22		12.40		1, 2		
L1498	04 07 50.0	25 02 13	(0, 0)	14.90		12.80		1		
L1495	04 10 57.5	28 03 13	(0, 0)	14.90		12.20		1		
L1521B	04 21 08.5	26 30 00	(0, 0)		13.15	13.20		1, 3		
L1400K	04 26 51.0	54 45 27	(0, 0)	14.60		12.10		1		
TMC-2A	04 28 58.4	24 24 27	(0, 0)	15.50	12.82			2, 4		
ТМС-2	04 29 43.0	24 18 54	(0, 0)	14.88	13.41	13.00	13.93	1, 2, 3, 4, 5, 6		
L1536B	04 30 21.7	22 38 10	(0, 0)	14.80	12.63	12.50		1, 3		
ТМС-3А	04 32 32.2	24 02 40	(0, 0)	14.75	12.77			5		
HCL2-B	04 36 47.0	25 58 15	(0, 0)	14.70	13.10	12.60	14.00	7		
TMC-1A	04 36 55.0	25 36 55	(0, 0)	14.90	13.00	12.60	14.30	7		
			(0, 5)	14.20	13.20	12.60	14.20	7		
			(0, 7.5)	14.20	13.30	12.60	14.30	7		
TMC-1C	04 38 32.0	25 55 00	(0, 0)	14.91	13.87	13.05	14.60	1, 5, 7		
			(2.5, -2.5)	14.80	13.00		•••	7		
			(1.25; -2.5)	14.60		12.20	14.00	7		
			(2.5, -1.25)	14.90	13.00			7		
			(1.25, -1.25)	15.10	13.60	13.00	14.50	7		
		(0, -1.25)	14.90	13.40	12.60	14.30	7			
		(2.5, 0)	14.90	13.20	•••		7			
			(1.25, 0)	15.00	13.40	12.90	14.20	7		
			(-1.25, 0)	14.80	13.40	12.80	14.10	7		
			(-2.5, 0)	14.90	13.20	12.30	•••	7		
			(0, 1.25)	14.90	13.30	12.60	14.20	7		
			(-1.25, 1.25)	14.90	13.20	12.50		7		
			(-2.5, 1.25)	14.80	13.50	12.80		7		
			(-2.5, 2.5)	•••		12.60	•••	7		
			(-1.25, 2.5)	14.40	•••	•••	•••	7		
			(0, 2.5)	14.60				7		
ТМС-1	04 38 42.0	25 35 45	(0, 0)		13.69	13.60	14.53	1, 2, 3, 6, 8		
			(-4, 6)	15.10			13.92	2, 6		
			(-2, 4)	14.91	•••		14.15	2, 6		
			(3.5, -4)	•••			14.26	2, 6		
			(1.67, 2.5)	•••	13.60	13.40	•••	8		
			(0.83, 1.25)	•••	13.60	13.40	•••	8		
			(-0.83, -1.25)	•••	13.80	13.40	•••	8		
1 16170	04 53 03 (20 22 19	(-1.0/, -2.5)		13.70	13.40	•••	8		
LISI/B	04 52 02.6	30 32 18	(0, 0)	14.80		12.40	•••	1, 2		
LI512	05 00 54.5	32 40 00	(0, 0)	14.90	12.45	12.90		1 2 2 5 6		
L1344	05 01 14.0	25 07 00	(0, 0)	15.55	13.45	13.40	13.81	1, 2, 3, 5, 6		
L134	15 50 53.0	-04 28 13	(0, 0)	•••	12.00	•••	14.39	0		
L1835	15 51 32.7	-02 42 51	(0, 0)	15.25	12.00	12.20	13.32	4, 5, 6		
L103	15 51 55./	-02 40 54	(0, 0)	13.33	12.91	12.30	•••	1, 3, 3		
L4JE	10 31 48.3	- 13 40 33	(0, 0)	14.00	12.57	12.40	12.40	116		
L05 I 778	10 4/ 21.0	- 18 01 00	(0, 0)	14.80	12.37	12.10	13.49	1, 4, 0		
L / /0	17 24 20.4 21 01 48 4	23 32 31 67 42 12	(0, 0)	14.00		12.50		1 1		
L1031B	21 01 40.0	47 18 17	(0, 0)	15.00	•••	12.50		1		
	21 TJ J2.0	7/ 1012	(0, 0)		•••	12.30		1		

^a Offsets are in arcminutes.

REFERENCES.—(1) Benson and Myers 1983; (2) Gaida, Ungerechts, and Winnewisser 1984; (3) Sorochenko, Tolmachev, and Winnewisser 1984; (4) Bujarrabal *et al.* 1981; (5) Churchwell, Nash, and Walmsley 1984; (6) Guelin, Friburg, and Mezaoui 1982; (7) Cernicharo, Guelin, and Askne 1984; (8) Tölle *et al.* 1981.

from each other by 0.2–0.4 dex and (2) different transitions were analyzed. Since we are mainly interested in the slopes of relationships, the actual scale for $N(C_4H)$ is of minor importance.

The values for $N(H_2)$ are derived from A_V using the relationship: $N(H_2) = 0.94 \times 10^{21} A_V$ cm⁻² (e.g., Myers, Linke, and Benson 1983). For many of the clouds, especially those in Taurus, the values for A_V are from Cernicharo and Bachiller (1984). Many of the sources for the molecular data tabulated A_V as well. The derivation of A_V involves uncertainties at the 15%-20% level, and when $A_V \ge 6$ mag, such a value is regarded as a lower limit (see, e.g., Cernicharo and Bachiller 1984). We limit ourselves to positions with $A_V \ge 1$ mag, where most of the gas is expected to be H₂, thereby allowing us to apply the above expression between $N(H_2)$ and A_V .

III. RELATIONSHIPS FOR OXYGEN-BEARING SPECIES

a) CO

The ultraviolet measurements of CO and H₂ in diffuse clouds reveal that N(CO) varies quadratically with $N(H_2)$ (Federman *et al.* 1980). Federman and Lambert (1988) showed that this trend continued in moderately reddened directions by plotting N(CO) versus N(CH); N(CH) was used because this

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molecule increases linearly with $N(H_2)$ for directions with $A_V \leq 4$ mag (Cardelli and Wallerstein 1986; Mattila 1986). Since the effects of self-shielding of the CO lines responsible for CO photodissociation occur at $N(CO) \geq 10^{16}$ cm⁻² (Glassgold, Huggins, and Langer 1985; van Dishoeck and Black 1988), the results for dark clouds are important for the CO chemistry. Results from the Bordeaux telescope (see Table 1) indicate that for dark clouds, $N(C^{18}O)$ is directly proportional to A_V or equivalently $N(H_2)$. There is considerable scatter in the data points for some of the clouds, however, and a nonlinear relationship may be more appropriate. We have investigated this possibility by compiling the data in a log-log plot so that we may derive the slope applicable to the kinetics of CO.

Figure 1 shows the results of the compilation for CO. Maps of entire dark clouds and selected observations of dark cloud cores in the $J = 1 \rightarrow 0$ line of C¹⁸O are used in the analysis (see Table 1). The C¹⁸O data are multiplied by 500, the isotopic ratio applicable to interestellar clouds (Wannier 1980), to obtain N(CO) from N(C¹⁸O). Selective photodestruction of C¹⁸O relative to CO may require the use of a slightly larger isotopic ratio (Bally and Langer 1982); the recent analysis of the photodestruction of carbon monoxide by van Dishoeck and Black (1988) indicates that the appropriate ratio may be closer to 1500. Increasing the ratio by ~3 would affect the display in Figure 1a slightly. The polygon representing the dark cloud would have to be raised 0.5 dex.

The presentation in Figure 1b, which concentrates on the dark cloud data, is discussed first. The data for several clouds, including cloud cores, show a strong correlation with a slope of ~ 1.5 . For several hundred data points, a correlation coefficient r of 0.78 is highly significant. However, the data for $N(H_2)$ in the cloud cores are probably lower limits because of the large extinctions associated with the cores— $A_V \ge 5-6$ mag. If greater values for $N(H_2)$ were more appropriate for these data, the data should be moved an unknown amount to the right, suggesting a flatter relationship for log $[N(H_2)] > 21.75$. A constant abundance is inferred from a slope of unity in plots

similar to Figure 1b. (In such plots, abundances can be derived because in cloud envelopes the two species occupy similar volumes.) A slope greater than unity indicates that the abundance of CO continues to increase for clouds with $N(H_2)$ between 1×10^{21} cm⁻² and 5–6 × 10²¹ cm⁻². This result contradicts the conclusions of Ungerer et al. (1985), Duvert, Cernicharo, and Baudry (1986), and Cernicharo and Guelin (1987); based on linear plots, these studies found no change in the abundance of C¹⁸O. An appreciable amount of dispersion, which could mask a relationship with a slope somewhat greater than 1, is seen in their linear plots. Any flattening of the relationship beyond $N(H_2) \sim 5 \times 10^{21}$ cm⁻² would suggest that C¹⁸O samples only the outer regions of a dark cloud, contrary to commonly held belief.

A comparison between CO and H_2 measurements of diffuse clouds and the measurements of $C^{18}O$ and A_V for dark clouds is illustrated in Figure 1a. As noted previously by Federman et al. (1980), the diffuse cloud data show a definite trend, and Figure 1a indicates that this trend is nearly continuous when the dark cloud data are included. This result bears on the models of the self-shielding of CO to photodissociation. In the case for H_2 , the ultraviolet data indicate a sharp (factor of 10^4) transition as the lines become optically thick to dissociating radiation (Savage et al. 1977). No sharp transition is apparent in our compilation for CO. Linear least-squares fits to the diffuse cloud data and the dark cloud data indicate a factor of 6 jump in N(CO) at $N(H_2)$ of 10^{21} cm⁻². An increase in the isotopic ratio between $\overline{C^{18}O}$ and CO only changes this conclusion slightly. The analysis of CO self-shielding by van Dishoeck and Black (1988) reveals that the sharpness of the transition depends on the incident flux of dissociating radiation. Their results suggest that a radiation field about half of the flux deduced for the local Galactic environment can reproduce the variation seen in Figure 1. A slight reduction caused by attenuation of the average Galactic radiation field in the diffuse envelope ($A_V \leq 1 \text{ mag}$) of a cloud seems appropriate.

Several points regarding the compilation of CO data require additional comment. Foremost, calculations for selective



FIG. 1.—A plot of log N(CO) vs. log $N(H_2)$. (a) Comparison of diffuse cloud data with dark cloud data. (b) Expanded view of dark cloud with the following sources: cores (circles) from Myers, Linke, and Benson (1983) and (triangles) from Baudry et al. (1981); (squares) for 3C 111–Ungerer et al. (1985); (inverted triangles) for L1495–Duvert, Cernicharo, and Baudry (1986); (diamonds) for HCL2–Cernicharo and Guelin (1987). Dashed line indicates a linear fit to the data with a slope of 1.6.

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photodissociation do not yet yield a quantitative value for the conversion factor from C¹⁸O to CO. The laboratory measurements on CO of Letzelter et al. (1987) were acquired at room temperature; thus, an unknown fraction of the predissociation seen by them may arise in absorption from rotational levels not populated at interstellar temperatures. van Dishoeck and Black (1988) acknowledge this uncertainty in their calculation of selective photodissociation and estimate the range of applicability to be a factor of 2. If at interstellar temperatures fewer transitions lead to photodissociation, an isotopic ratio approaching the terrestrial value of 500, the value used by us, would be appropriate. Second, should the isotopic ratio be larger than the terrestrial value, Figure 11 of van Dishoeck and Black (1988) indicates an increasing value in C¹⁸O/CO for A_{ν} between 2 and 4 mag. The factor of 2 increase in the ratio $C^{18}O/CO$, comparable to the largest uncertainties expected in the analysis, would raise the data with values of $\log [N(H_2)] \sim$ 21.30 preferentially. The result would be a slope approaching 2 for smaller values of $N(H_2)$ and a slope somewhat less than 1.5 at larger values of $N(H_2)$. Third, the theoretical curves of van Dishoeck and Black reveal a nonlinear relationship between N(CO) and $N(H_2)$, a result that conflicts with the linear correlations found by Ungerer et al. (1985), Duvert, Cernicharo, and Baudry (1986), and Cernicharo and Guelin (1987). Our log-log presentation indicates a slope of ~ 1.5 , a value more in line with the theoretical results. Last, a slope of ~ 1.5 indicates that the abundance of CO continues to increase for molecular material with A_V of 1–4 mag. The increase in CO abundance occurs in the same portions of a cloud that optical observations show a decrease in C^+ abundance (e.g., Federman and Huntress 1989). The carbon ion is being chemically processed into the stable CO molecule.

We have not included the results of Frerking, Langer, and Wilson (1982) because of the potential problem with their comparisons between $N(C^{18}O)$ and A_V (see Cernicharo and Guelin 1987). van Dishoeck and Black (1988) suggested that the two sets of observations can be reconciled by allowing somewhat larger uncertainties in the measured parameters. In any case, the addition of the small number of data from Frerking, Langer, and Wilson would not alter our findings.

More important to our analysis are some of the conclusions of Frerking, Langer, and Wilson (1982). Their results show (1) direct evidence for self-shielding through a break in the relationships between $N(C^{18}O)$ or $N(^{13}CO)$ and A_V [or $N(H_2)$] at a column density of ~ 10^{14} cm⁻² and (2) a nonlinear relationship with A_V because of the break. The break is an indication of where self-shielding is playing a role. A break in the relationship for $N(C^{18}O)$ at ~ 10^{14} cm⁻² corresponds to a break in the relationship for N(CO) at ~ 10^{16} cm⁻², as found here. However, their conclusions and those of Bally and Langer (1982) refer to breaks for all isotopic variants at $\sim 10^{14}$ cm^{-2} . We do not see a break in our compilation until N(CO)reaches a value of $\sim 10^{16}$ - 10^{17} cm⁻², consistent with the results of van Dishoeck and Black (1988). The relatively tight correspondence between N(CO) and $N(H_2)$ for column densities less than 10^{16} cm⁻² merits a closer theoretical examination of the self-shielding process. Frerking, Langer, and Wilson also found indications of saturation in the C¹⁸O lines above extinctions of ~ 6 mag, a result implying that the relationship between N(CO) and $N(H_2)$ remains nonlinear beyond log $[N(H_2)] \sim 21.75$, A nonlinear relationship cannot continue much beyond this column density in H₂, however, because the abundance of CO is not likely to exceed a value of $\sim 10^{-4}$.



FIG. 2.—Compilation of 6 cm H₂CO data plotted against $N(H_2)$. The sources for the data are (*circles*) Pöppel, Rohlfs, and Celnik (1983)–TMC-2; (*squares*) Minn and Greenberg (1979)–globules near NGC 2264; (*diamonds*) Sherwood and Wilson (1981)–HCL2; (*triangles*) Cohen *et al.* (1983)–various sources; (*inverted triangles*) Minn and Greenberg (1987)–L1709. Two linear fits are shown: the thin line with a slope of 2.33 based on the data of Pöppel, Rohlfs, and Celnik and the dashed line with a slope of 1.62 based on all the data.

b) H₂CO

Our compilation of 6 cm H₂CO observations is presented in Figure 2; the references appear in Table 1. Data for the ρ Oph Cloud (e.g., Myers *et al.* 1978) and for the ϵ Cha I dark cloud (Toriseva and Mattila 1985) are not included because these data sample regions where significant amounts of star formation are taking place. Such data fall below the data shown in Figure 2 by ~0.3 dex, even for A_V approaching 4 mag, indicating that enhanced ultraviolet flux from the nearby star formation readily dissociates H₂CO. We have placed the highest weight on the data of Pöppel, Rohlfs, and Celnik (1983) because of the number of data points in their study of TMC-2 (L1529) and because they used the 100 m telescope at Bonn which has the highest spatial resolution available at 6 cm.

The data of Pöppel, Rohlfs, and Celnik (1983) show a correlation with $N(H_2)$. The slope of the relationship is 2.33; the correlation is quite strong with r equaling 0.70 for 107 data points. The combined set of data reveals a slightly smaller slope of 1.62, but a similar degree of correlation (r = 0.69 for over 200 data points). A synthesis of these two results suggests that $N(H_2CO)$ varies approximately in a quadratic manner with $N(H_2)$. Different relationships are found from individual H_2CO studies when the data are displayed on linear axes. For example, no correlation was seen by Minn and Greenberg (1979) for globules near NGC 2264, but a correlation indicating direct proportionality appeared in the data of Sherwood and Wilson (1981) for HCL-2. As discussed in the introduction, difficulties arise with such analyses. The compiled set of data, therefore, are graphed in a log-log fashion.

Photodissociation of H₂CO is likely to be important for visual extinctions approaching 4 mag because H₂CO, which has two relatively weak hydrogen bonds, is readily destroyed by photons with wavelengths greater than 1200–1300 Å (Suto, Wang, and Lee 1986). This fact translates into a thicker region of photodissociation for H₂CO compared to the region appropriate for the more tightly bound molecules C₂ and CN, which are dissociated by photons with $\lambda \leq 1200-1300$ Å (Pouilly *et*

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al. 1983 for C₂; Lavendy, Robbe, and Gandara 1987 for CN). Since H₂CO is destroyed by redder photons than are C₂ or CN, and since the amount of extinction increases with a decrease in wavelength (see Code *et al.* 1976), the photochemical regime of H₂CO penetrates deeper into a cloud. Moreover, the fact that gas phase reactions are less efficient at destroying H₂CO enhances the importance of photodissociation. For example, the reaction involving H₂CO and O occurs at a relatively slow rate (e.g., Washeda, Martinez, and Bayes 1974). (Ion-molecule reactions tend to recycle formaldehyde.)

For the highest values of $N(H_2)$, which as mentioned above for CO are probably lower limits, the slope between log $N(H_2CO)$ and log $N(H_2)$ may actually decrease because the relationship levels off. The likelihood of sampling the portion of cloud where H_2CO is photodissociated and the possibility of a decreasing slope indicate that 6 cm measurements of H_2CO always arise from the outer regions of dark clouds. This conclusion was reached previously by Vanden Bout, Snell, and Wilson (1983), who partially mapped several dark clouds in the 6 cm transition. Minn and Greenberg (1987) suggested that depletion onto grains is the cause of the decrease in slope as observations probe the core of the cloud.

Finally, we want to stress that the observational data presented in Figure 2 are in disagreement with pure gas-phase photochemical equilibrium models. The theoretical estimates are significantly lower than the observed column densities. We hope that this compilation will be useful in improving our theoretical understanding.

IV. RELATIONSHIPS FOR NITROGEN-BEARING MOLECULES

a) NH_3

Ammonia is one of the simplest nitrogen-bearing species predicted to be abundant in dark clouds; readily detectable lines at radio wavelengths make NH₃ a species commonly observed in molecular clouds. The excitation conditions for emission of NH₃ (densities $\sim 10^4$ cm⁻³) occur in the core regions of dark clouds (Benson and Myers 1983). Thus $N(NH_3)$ is not expected to correlate with A_V or $N(H_2)$, which sample both high- and low-density regions within a cloud. This fact is borne out in our compilation. A more appropriate molecule to compare with NH₃ may be C¹⁸O, which because of its low isotopic abundance, samples deeper into the cloud than does CO. Figure 3 shows the correspondence between $N(NH_3)$ and $N(C^{18}O)$. A somewhat surprising result first noted by Benson and Myers (1983) is that there is no correlation between these two molecules. One possible reason for the lack of a correlation may be that the NH₃ only resides in clumps within the core.

Millar and Freeman (1984*a*, *b*) offered the suggestion that reactions involving atomic nitrogen lead to the cyanopolyynes. If atomic nitrogen is also involved in the production of NH_3 , as was thought to be the case in earlier models (Millar and Freeman 1984*a*), then we expect some correlation between ammonia and the cyanopolyynes. We investigate this by plotting log $N(HC_3N)$ and log $N(HC_5N)$ against log $N(NH_3)$ in Figures 4 and 5. No correlation is seen in either plot, implying that reactions such as

$$H_3^+ + N \to NH_2^+ + H$$
, (5)

are not important in the production of NH_3 . We interpret this result as an astronomical substantiation of previous analyses



FIG. 3.—Comparison of log $N(NH_3)$ and log $N(C^{18}O)$. The sources are (circles) Benson and Myers (1983) in combination with the other references listed in Table 1 for cores and (triangles) Cernicharo, Guelin, and Askne (1984) and Gaida, Ungerechts, and Winnewisser (1984) for clouds in Taurus.

of the reaction dynamics of equation (5) (Huntress 1977; Herbst, Defrees, and McLean 1987). When the fine structure of N^+ is taken into account (Galloway and Herbst 1989), or when the kinetic energy of N^+ from the dissociative charge transfer of He⁺ on N₂ is considered (Yee, Lepp, and Dalgarno 1987), the reaction

$$N^+ + H_2 \rightarrow NH^+ + H , \qquad (6)$$

is sufficiently rapid nearly to produce the observed amount of NH_3 (Herbst and Leung 1989). Langer and Graedel (1989) also found that the abundance of NH_3 relative to H_2 in TMC-1 can be reproduced by incorporating reactions between CH_3^+ and N.

An important fact is revealed in Figures 3-5, especially the



FIG. 4.—Same as Fig. 3 for HC_3N and NH_3 . (*triangles*): from Cernicharo, Guelin, and Askne (1984) for Taurus sources. (*circles*): from a combination of Bujarrabal *et al.* (1981), Benson and Myers (1983), Churchwell, Nash, and Walmsley (1984), Gaida, Ungerechts, and Winnewisser (1984), and Sorochenko, Tolmachev, and Winnewisser (1984).



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FIG. 5.—Same as Fig. 3 for HC_5N and NH_3 . Symbols represent the same data sources as in Fig. 4.

latter two: log $N(NH_3)$ is always 15! While the other molecules studied here show systematic variations in column density of factors of 10–100, the column density of NH₃ is constant to within a factor of 3. The constancy in $N(NH_3)$ may be due to unusual chemical processes or excitation conditions for NH₃. At the present time, however, we place more emphasis on a better understanding of the excitation conditions.

Maps of ammonia and cyanopolyyne emission from TMC-1 reveal distinctly different distributions (Tölle *et al.* 1981; Olano, Walmsley, and Wilson 1988). The general conclusion has been that real abundance variations are present, although Bujarrabal *et al.* (1981) interpreted the differences in terms of differences in excitation. Our finding that $N(NH_3) \sim 10^{15}$ cm⁻², with little dispersion about this value for an ensemble of clouds, may hold the clue to resolving this issue.

b) Complex Molecules: HC₃N, HC₅N, and C₄H

Most comparisons between complex molecules have been restricted to two dark clouds, TMC-1 and L183 (L134N). Here we incorporate appropriate data for as many clouds as possible, especially the surveys of dark cloud cores of Myers and collaborators. The comparison of $N(\text{HC}_5\text{N})$ with $N(\text{HC}_3\text{N})$ is shown in Figure 6. For this compilation, two points are worthy of mention. First, a strong correlation with a slope of 1 and a correlation coefficient of 0.78 exists between these molecules for all the cores. This result indicates that, as expected, the chemistries of these two molecules are connected. Second, *the data show that for all dark cloud cores, not just TMC-1*, $N(\text{HC}_3\text{N}) \sim 3 N(\text{HC}_5\text{N})$, a result noted previously by Bujarrabal *et al.* (1981) and Snell *et al.* (1981) for a smaller sample. Such a "universal" ratio places constraints on the mechanisms for building the carbon chains.

As can be discerned from Figure 7, a possible correlation exists between HC_5N and $C^{18}O$. The correlation coefficient is small, however ($r \sim 0.40$). Benson and Myers (1983) obtained a similar result for these data and attributed it to two distinct populations of clouds: those in Taurus and those elsewhere. We believe that the weakness of the correlation most likely arises because the two molecules are not chemically related, nor are they spatially coexistent. Since the data for HC_5N sampled the dense, core region, the weak correlation indicates



FIG. 6.—Same as Fig. 3 for HC_5N and HC_3N . The following sources of data were used: (*circles*) Benson and Myers (1983) in combination with Churchwell, Nash, and Walmsley (1984); (*triangles*) Cernicharo, Guelin, and Askne (1984); (*squares*) Tölle *et al.* (1981); (*diamonds*) Benson and Myers in combination with Gaida, Ungerechts, and Winnewisser (1984). The thin line indicates a least-squares linear fit to the data with a slope of 1.

that the observed C¹⁸O may not reside in the core. This result is consistent with a decrease in slope at large $N(H_2)$ in Figure 1b.

The processes for producing the long-chain molecule C_4H are expected to be similar to those for the cyanopolyynes. The results of our compilation for $N(C_4H)$ are plotted against $N(HC_5N)$ in Figure 8; a similar correspondence is found between C_4H and HC_3N . Again, a correlation is present; the slope is approximately 0.5, and r is about 0.80. The correlation was first noted by Millar and Freeman (1984a). The correspondence between carbon-chain molecules both with and without nitrogen leads to the suggestion that all long-chain molecules are formed through similar processes; for example, reactions involving acetylene can build the backbone of these molecules



FIG. 7.—Same as Fig. 3 for HC_5N and $C^{18}O$, where the data are from Myers and collaborators.



FIG. 8.—Same as Fig. 3 for C₄H and HC₅N. The data of Guelin, Friberg, and Mezaoui (1982) (circles) have been scaled to the data of Cernicharo, Guelin, and Askne (1984) (triangles).

(Mitchell and Huntress 1979; Schiff and Bohme 1979). More data are needed to substantiate this finding.

The results of Snell et al. (1981) have not been included in our compilation, primarily for two reasons. Their data for L134N (L183) were acquired at a different position than the data of Myers and collaborators, and their column densities are significantly lower for many of the clouds in common with the more recent work presented above. A large part of the differences in column density probably resides in the fact that Snell et al. derived $N(HC_3N)$ from physical conditions, which were obtained by analyzing HC₅N emission, assuming optically thin emission from HC₃N, and applying a large velocity gradient model of radiative transfer to the data. The correspondence between HC₅N and HC₃N that we found in Figure 6 probably arises from chemical ties and need not indicate spatial coexistence. Other observational results, based in part on analyzing the emission of hyperfine components, found moderate optical depths for HC₃N emission (e.g., Cernicharo, Guelin, and Askne 1984; Sorochenko, Tolmachev, and Winnewisser 1986). Thus, Snell et al. may have underestimated $N(HC_3N)$ through an incorrect assessment of density and through a neglect of line saturation.

V. SUMMARY

We analyzed the chemistry of dark clouds from an empirical perspective. Correlations among measured column densities were sought. The slopes arising from log-log plots were determined because these slopes provide constraints for the chemistry of interstellar molecules and for the physical structure of the cloud.

We find that when C¹⁸O data are converted to CO through

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the use of a constant ratio of 500, the slope in the relationship between CO column density and H₂ column density in clouds with $A_V \leq 4$ mag is ~1.5. This result indicates that the abundance of CO increases with depth in the outer regions of dark clouds, but less steeply with $N(H_2)$ than in diffuse clouds with $A_V \leq 1$ mag. A factor of 6 jump occurs at $N(H_2)$ of 10^{21} cm⁻² between the column density for CO in dark clouds and that in diffuse clouds, a result that places constraints on the strength of the UV field responsible for the photodissociation of CO. Measurements of the 6 cm line of H_2CO indicate that $N(H_2CO)$ increases quadratically with $N(H_2)$ for $A_V \le 4$ mag, but that the relationship probably flattens for data appropriate to $A_V \ge 4$ mag, where only lower limits for $N(H_2)$ are available.

The data on nitrogen compounds and carbon-chain molecules discussed in § IV also reveal information. $N(NH_3)$ is a nearly constant value of $\sim 10^{15}$ cm⁻² for all clouds sampled. Because of recent progress in the understanding of ammonia chemistry, a cause for the constant column density may arise from excitation. There is a highly significant correlation between $N(HC_5N)$ and $N(HC_3N)$ with a slope of unity. This relationship strongly suggests that the chemistry of these two species is connected. Moreover, the data for several dark clouds indicates that the value of ~ 3 for the ratio $N(\text{HC}_3\text{N})/N(\text{HC}_5\text{N})$ is universal, i.e., not specific to TMC-1. The weakness of the correlation between the cyanopolyynes and C¹⁸O favors the conclusion that the molecules are spatially distinct. Although the amount of appropriate data is sparse, the cyanopolyynes also appear correlated with C₄H, from which we infer a common origin for carbon-chain molecules with acetylene playing a crucial role.

We have not discussed several other important molecules that are relatively easily observed; these include CH, OH, and C_2 H. Mattila (1986) analyzed in detail the 9 cm observations of CH in dark clouds. More ultraviolet data from diffuse clouds and more 18 cm data from dark clouds taken at high spatial resolution are needed to study OH. Wootten et al. (1980) partially mapped several dark clouds in lines of C₂H; additional data are required before definitive statements regarding trends can be made.

These results add essential ingredients to models of the chemical and physical conditions in interstellar clouds. Knowledge of how strongly the column density of one molecule correlates with that for another molecule and knowledge of the slope and the intercept in the logarithmic relation provide clues to the underlying chemistry for the molecules and to the physical model for the cloud.

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